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Modeling N₂O production by ammonia oxidizing bacteria: single pathway or two-pathway models?

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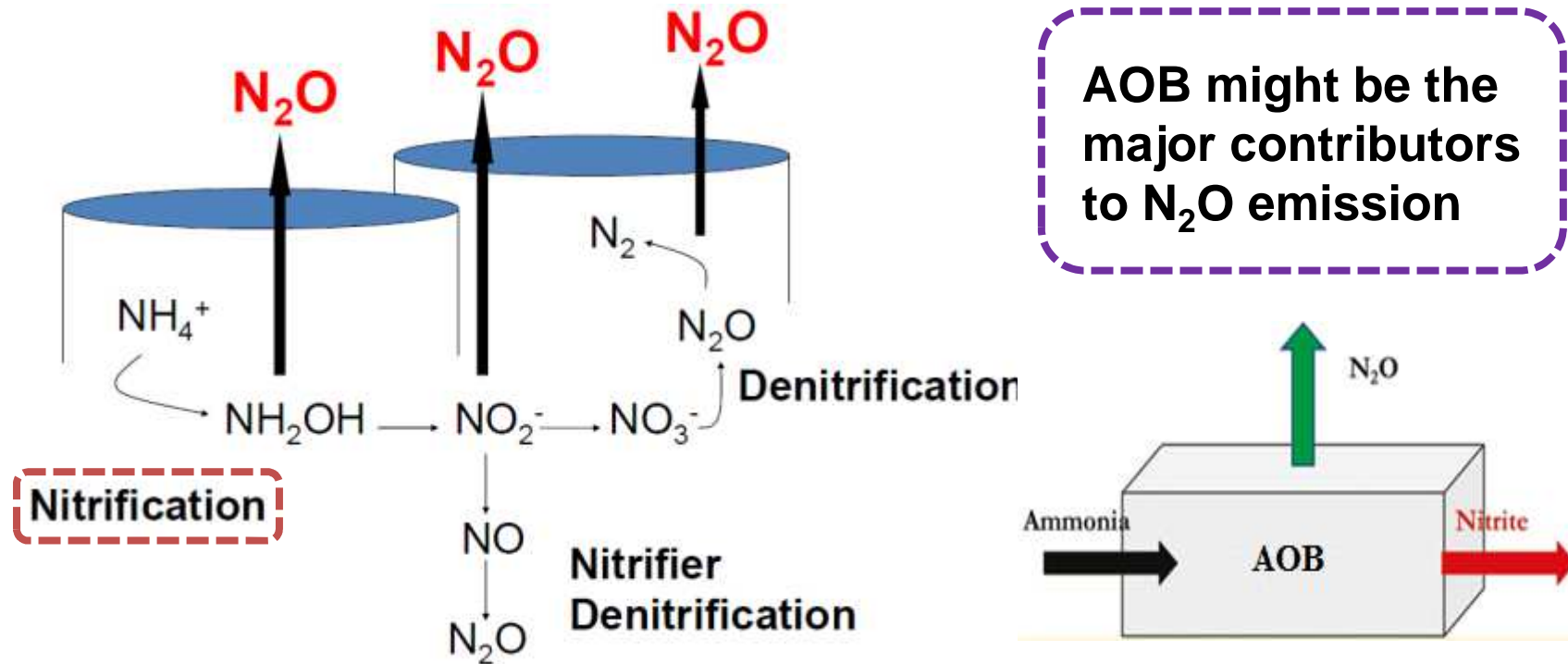
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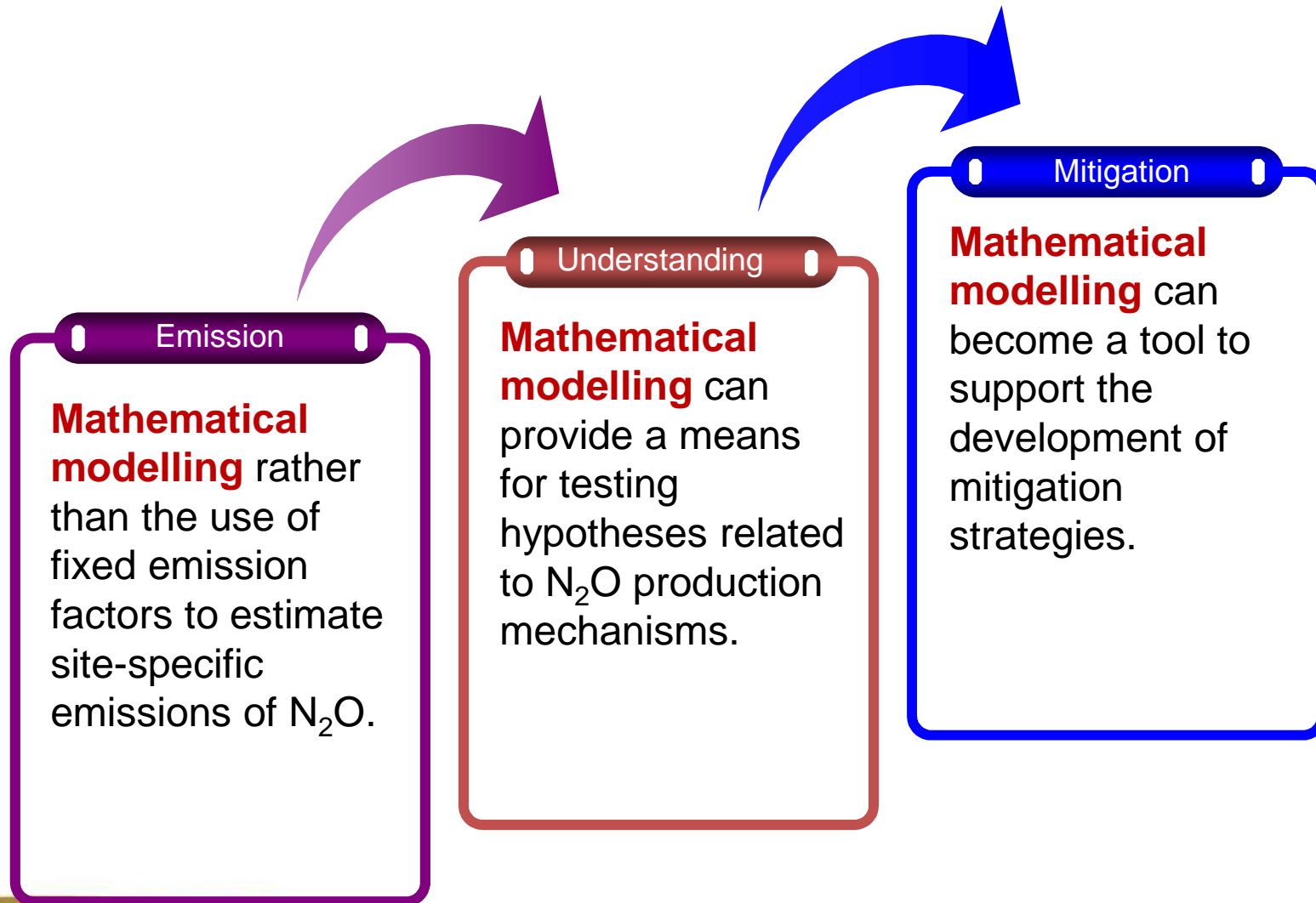


N₂O production processes

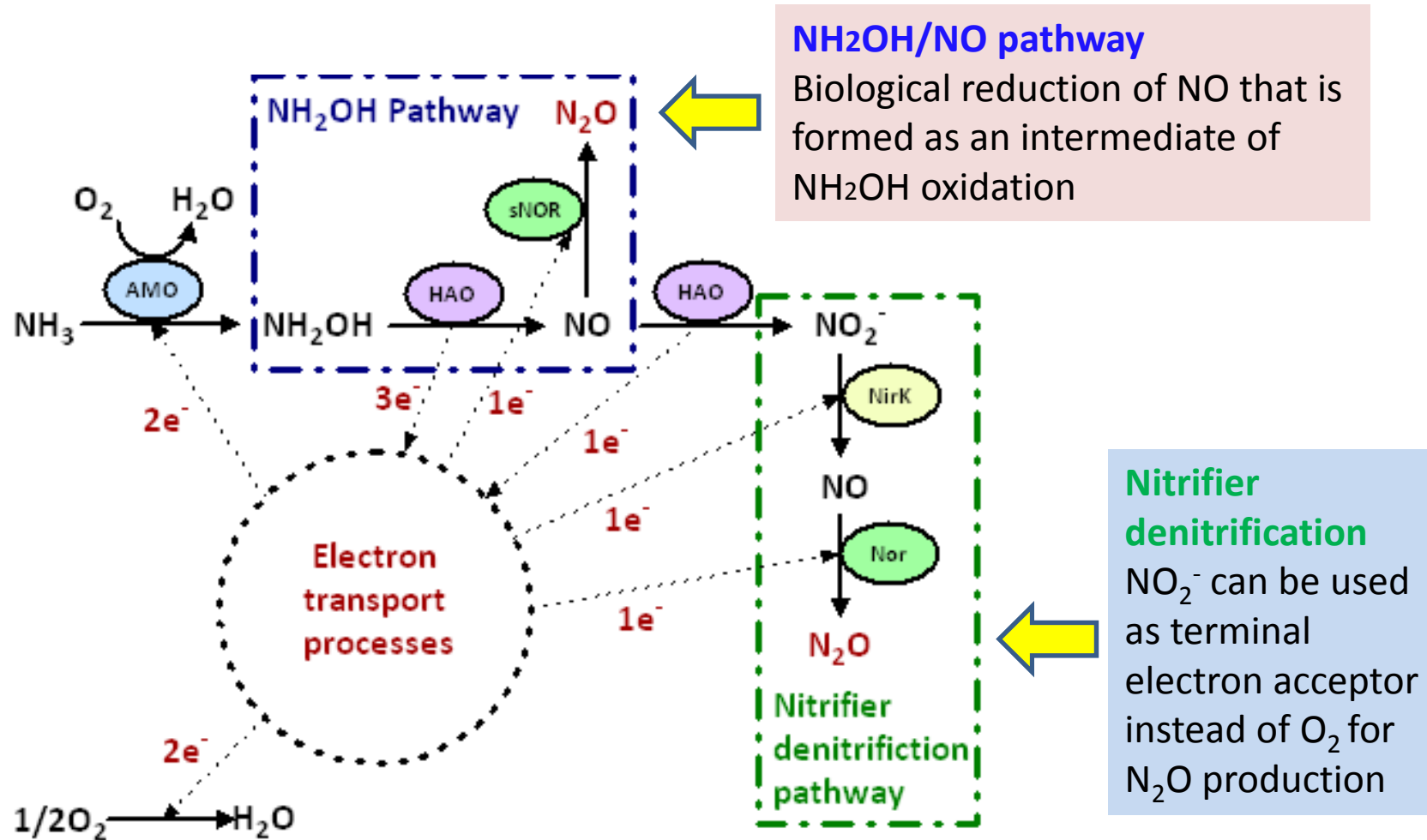


N₂O is produced during biological nitrogen removal typically attributed to **autotrophic ammonia-oxidizing bacteria (AOB)** and heterotrophic denitrifying organisms.

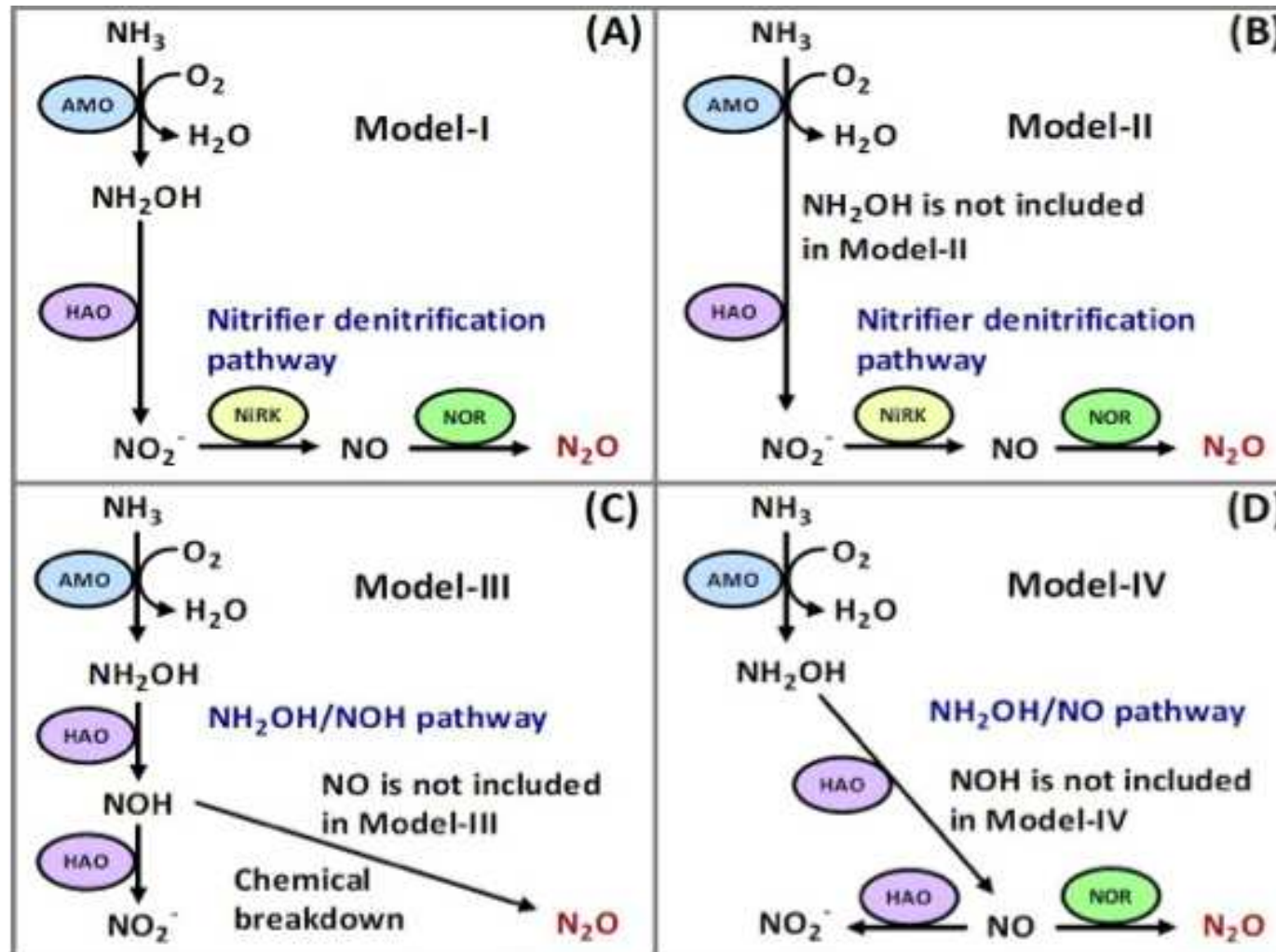
Mathematical Modelling of N₂O Production



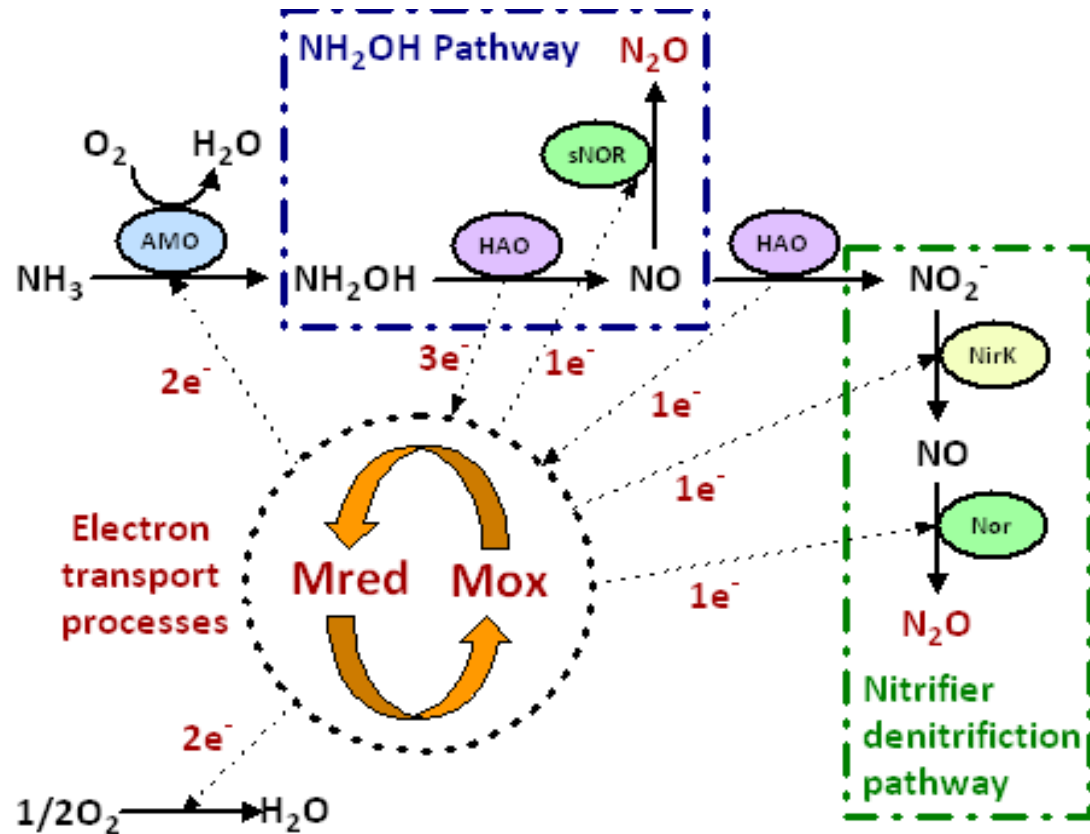
N₂O production pathways by AOB



Single-pathway models for N₂O production by AOB



Two-pathway model for N₂O production by AOB



Mred: electron carriers in reduced form
Mox: electron carriers in oxidised form

- decoupling the oxidation and reduction processes
- electron carriers as new model components
- total level of electron carriers being constant
- electron distribution between O₂, NO₂⁻ and NO with different kinetics controlling the **relative contributions between the two N₂O pathways**

Single pathway vs. two-pathway models

Two-pathway model:

- advantages: applicable under varying conditions
- disadvantages: complicated with more parameters

These applicable regions have not been identified!!!

Single-pathway

- ✓ advantages: simple
- ✓ disadvantages: applicable under certain conditions



d fewer parameters
ertain conditions

Methodology – model evaluations

Two-pathway model generate simulation data for calibration of single-pathway models

- **Steady-state calibration:**

- 1) calibrated η_{AOB} is constant, independent of the DO and nitrite levels;

- 2) η_{AOB} is in a feasible range (0 – 1.0) under any conditions.

- **Dynamic verification:**

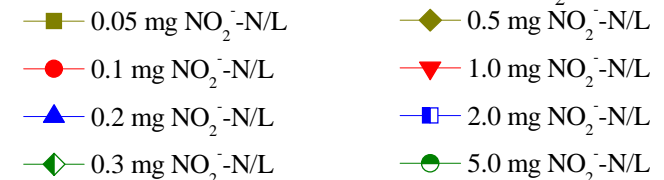
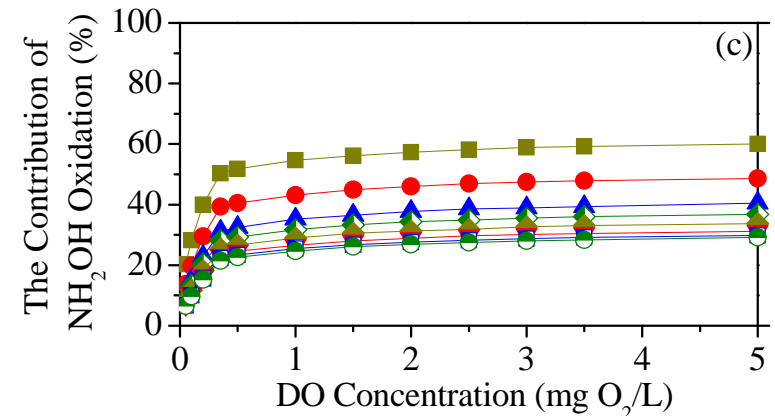
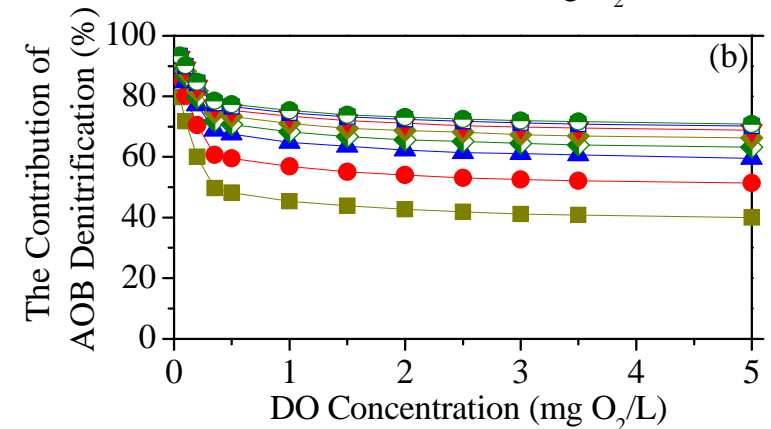
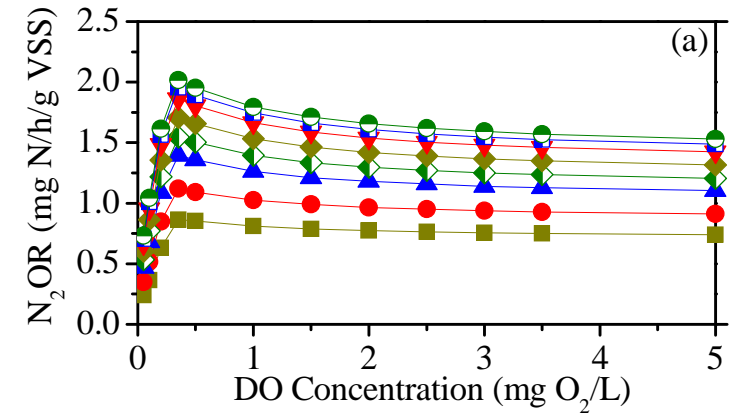
- 1) $F(p) = (\sum_{i=1}^n (y_{TM,i} - y(\eta_{AOB})_i)^2)^{\frac{1}{2}}$

- 2) ANOVA to evaluate the fit

N₂O production by AOB

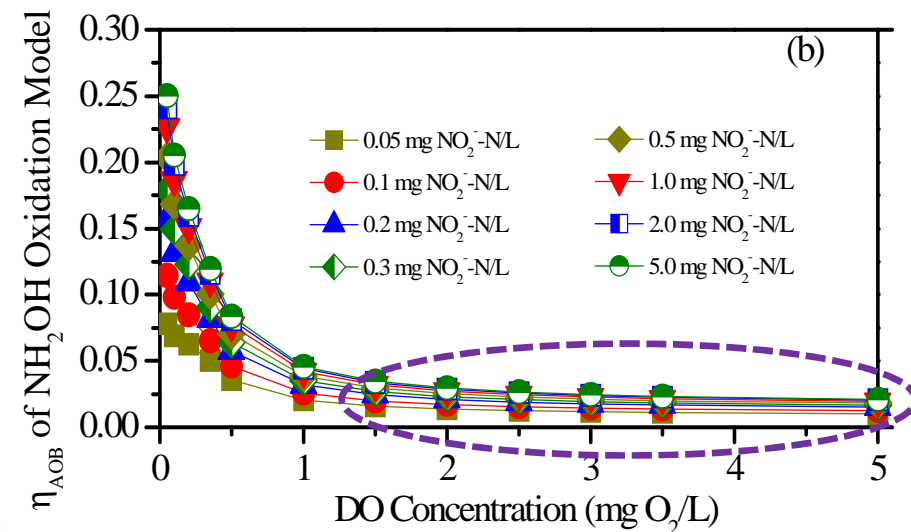
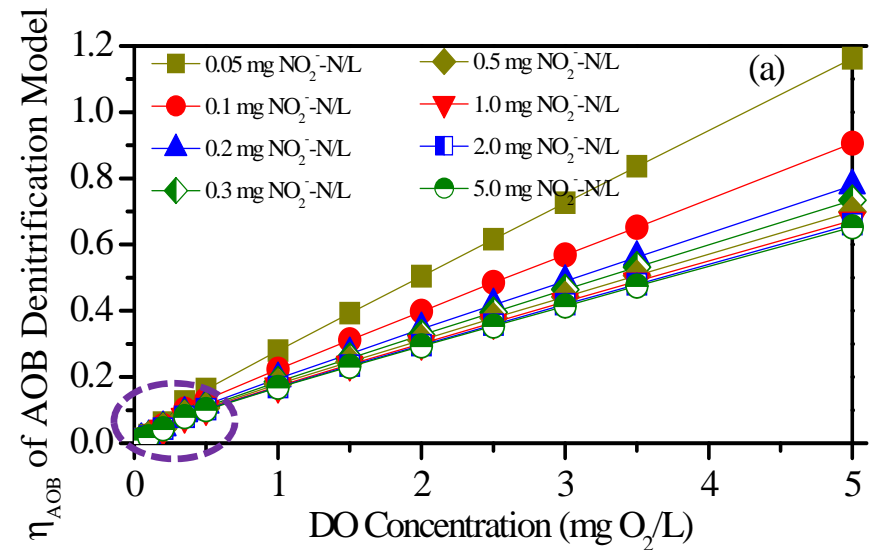
two pathway model

- ✓ Biomass specific N₂OR increases rapidly as DO increases from 0.05 to 0.35 mg O₂/L, and then decreases as DO further increases to 5.0 mg O₂/L.
- ✓ At each DO concentration, the simulated biomass specific N₂OR increases as NO₂⁻ increases.
- ✓ **The contribution of AOB denitrification pathway decreases with the increase of DO, but increases with the increase of NO₂⁻, accompanied by opposite changes in the contribution of NH₂OH oxidation pathway.**



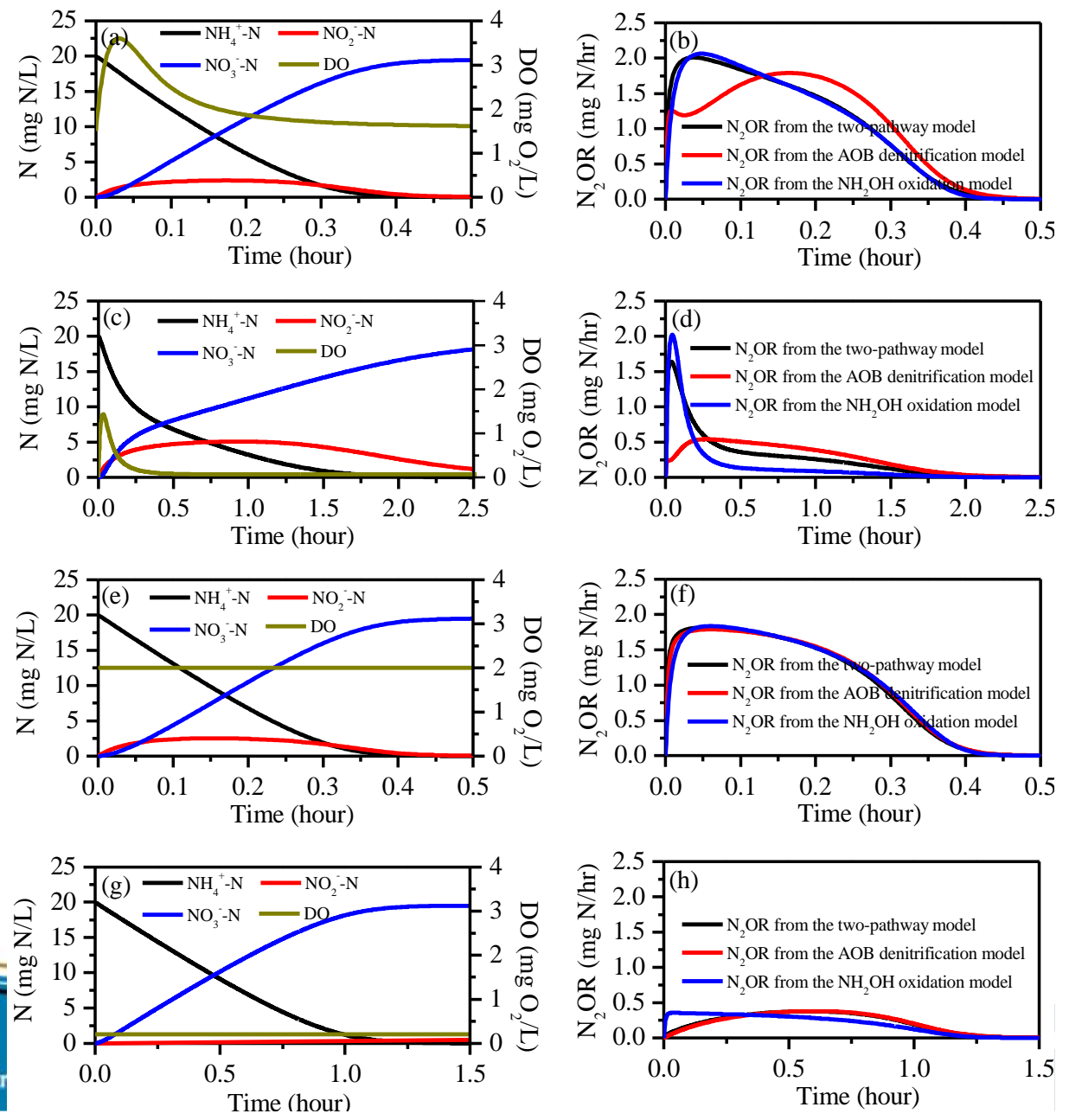
Steady-state calibration of single-pathway model

- ✓ The AOB denitrification model would only be applicable when DO is constant: in the range of 0.5 – 5 mg O₂/L, while nitrite varies in the range 1.0 – 5.0 mg NO₂⁻-N/L; in the DO range of 0 – 0.5 mg O₂/L, independent of the nitrite concentration
- ✓ NH₂OH oxidation model can be applied at high DO concentrations (> 1.5 mg O₂/L), which is not a suitable model under low DO conditions (<1.5 mg O₂/L).



Dynamic verification of the steady-state results

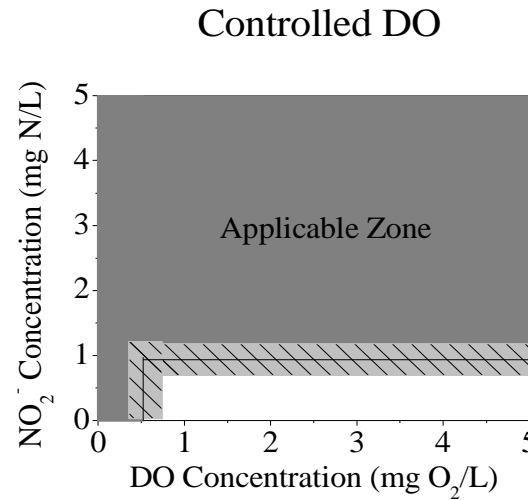
- DO varies between 1.5 – 3.5 mg O₂/L and NO₂⁻ accumulates to around 2.5 mg N/L.
- DO varies between 0 – 1.5 mg O₂/L and NO₂⁻ accumulates to approximately 5 mg N/L.
- Constant DO of 2.0 mg O₂/L and nitrite accumulation of 2.5 mg N/L
- Constant DO (0.2 mg O₂/L) and without nitrite accumulation (below 0.5 mg N/L)



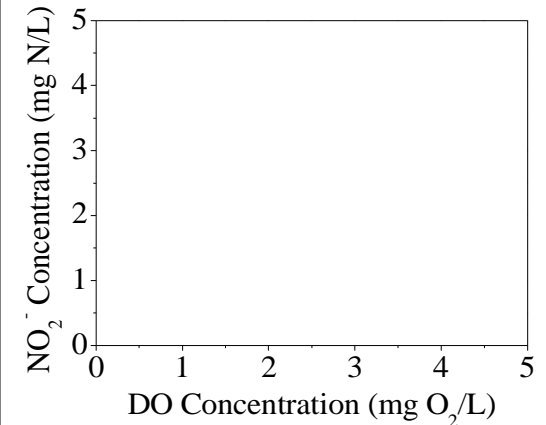
Effect of model parameter of two pathway model?

The applicable regions were generally insensitive (below $\pm 10\%$) to the variations of key parameter values by $\pm 50\%$ of the two-pathway model

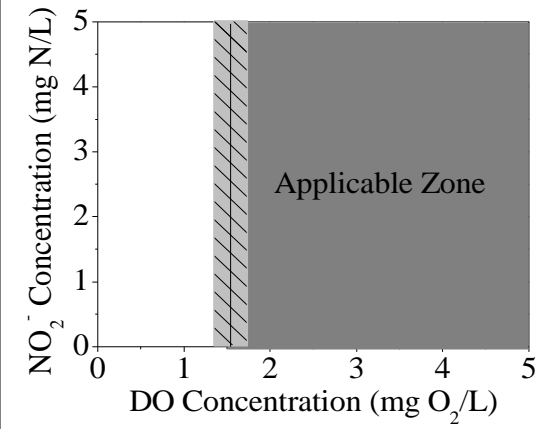
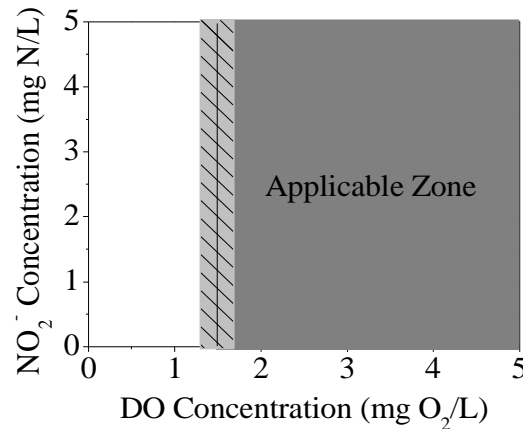
AOB Denitrification Model



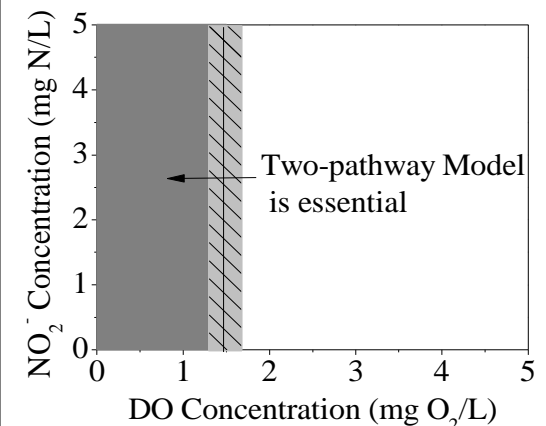
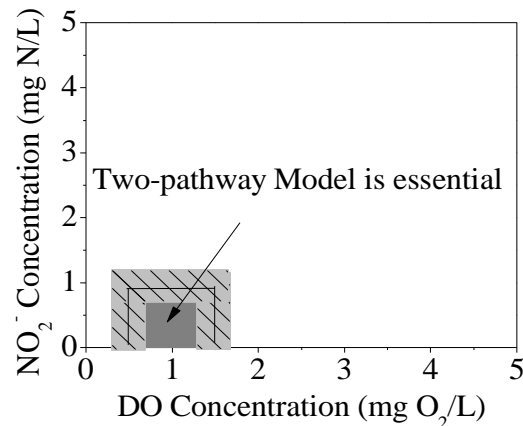
Dynamic DO



NH2OH Oxidation Model



Two-pathway Model



Key messages of this work

- AOB denitrification model: DO is well controlled at a constant level, either at low DO concentration (<0.5 mg O₂/L) with any investigated NO₂⁻ concentration (0 – 5 mg N/L) or at high DO (≥ 0.5 mg O₂/L) with significant NO₂⁻ accumulation (≥ 1.0 mg N/L).
- NH₂OH oxidation model can be applied under high DO conditions (≥ 1.5 mg O₂/L), controlled or varying, with any NO₂⁻ concentration investigated (0 – 5.0 mg N/L).
- Under other conditions, the two-pathway model should be applied.

Thank you!



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